

- Typical NMR tube diameter: 5 mm
- Sample volume: > 600 μL
- Smaller volume (200-300 μL) in Shigemi tubes
- Concentration for structure determination: several hundred μM
- Lower concentration for ligand binding
- Signal to noise ratio (S/N) $\alpha \sqrt{n}$
- Cryoprobe: electronics cooled to ~ 20K, higher
 S/N (up to 4 fold), lower concentrations possible
- Higher S/N (& better resolution) at higher magnetic field
- Any buffer but preferably salt < 200 mM, pH < 7.5
- For homonuclear experiments, phosphate buffer or deuterated buffer
- 5-10% deuterated solvent (²H₂O) for lock

2-dimensional (2D) NMR spectroscopy

Correlation between nuclear dipoles

- Through-bond interactions (J-coupling) < 4 bonds apart
- Through-space interactions (nuclear Overhauser effect, NOE) < 5 Å apart



Common 2D homonuclear (¹H) NMR experiments

- Useful for proteins up to ~ 100 residues
- Also useful for small(er) molecules

COSY (COrrelation SpectroscopY): crosspeaks between J-coupled spins (2 or 3 bonds apart)

TOCSY (Total Correlation SpectroscopY): crosspeaks within J-coupled networks (spin systems)

NOESY (NOE SpectroscopY): crosspeaks between spins close in space (< 5Å apart)

1D NMR spectrum



Cavanagh, J., Fairbrother, W.J., Palmer, A.G.III, Rance, M., Skelton, N.J. Protein NMR Spectroscopy: Principles and Practice, 2nd edition, 2007, Academic Press

TABLE 1 RANDOM COIL 'H CHEMICAL SHIFTS FOR THE 20 COMMON AMINO ACIDS WHEN FOLLOWED BY ALANINE

Residue	NH	Hα	H^{β}	Others
Ala	8.24	4.32	1.39	
Cys (reduced)	8.32	4.55	2.93, 2.93	
Cys (oxidized)	8.43	4.71	3.25, 2.99	
Asp	8.34	4.64	2.72, 2.65	
Glu	8.42	4.35	2.06, 1.96	yCH, 2.31, 2.31
Phe	8,30	4.62	3.14, 3.04	2,6H 7.28; 3.5H 7.38; 4H 7.32
Gly	8.33	3.96		
His	8.42	4.73	3.29, 3.16	2H 8.58; 4H 7.29
fle	8.00	4.17	1.87	γCH ₂ 1.45, 1.16; γCH ₂ 0.91; δCH ₂ 0.86
Lys	8.29	4.32	1.84, 1.75	γCH ₂ 1.44, 1.44; δCH ₂ 1.68, 1.68; εCH ₂ 2.99, 2.99; εNH; 7.81
Leu	8.16	4.34	1.62, 1.62	γCH 1.59: δCH, 0.92, 0.87
Met	8.28	4,48	2.11, 2.01	γCH_{2} 2.60, 2.54; ϵCH_{2} 2.10
Asn	8.40	4.74	2.83. 2.75	yNH, 7.59, 6.91
Pro	-	4.42	2.29, 1.94	yCH, 2.02, 2.02; &CH, 3.63, 3.63
Gln	8.32	4.34	2.12, 1.99	vCH, 2.36, 2.36; 8NH, 7.52, 6.85
Arg	8.23	4.34	1.86, 1.76	YCH, 163, 163, 6CH, 320, 320; ENH 807
Scr	8.31	4.47	3.89. 3.87	
Thr	8.15	4.35	4.24	vCH, 1.21
Val	8.03	4.12	2.08	vCH, 0.94, 0.93
Trp ^a	8.25	4.66	3.29. 3.27	2H 7 27: 4H 7 55: 5H 7 18: 6H 7 25: 7H 7 50
Tyr	8.12	4.55	3.03, 2.98	2,6H 7.14; 3,5H 6.84

Chemical shifts are referenced to internal DSS at 25 °C, pH ~5.0.

* Measured using a peptide with free N- and C-termini.

Wishart DS, Bigam CG, Holm A, Hodges RS, Sykes BD. 1H, 13C and 15N random coil NMR chemical shifts of the common amino acids. I. Investigations of nearest-neighbor effects. J Biomol NMR. 1995 Jan;5(1):67-81.

COSY (COrrelation SpectroscopY): crosspeaks between J-coupled spins • TOCSY (Total Correlation SpectroscopY): crosspeaks within J-coupled networks •+•



NOESY (NOE SpectroscopY): crosspeaks between spins close in space (< 5Å apart)

- sequential assignment: connect sequential spin systems (find neighbors)
- structural information



Wüthrich, K. NMR of Proteins and Nucleic Acids, 1986, John Wiley & Sons

NOESY + O COSY+TOCSY •





Common 2D heteronuclear NMR experiments

- Useful for both small and large proteins
- Typically performed on proteins labeled with ¹⁵N and/or ¹³C
- To incorporate ¹⁵N or ¹³C into proteins, the proteins are expressed in *E. coli* or yeast in a growth medium that only contains ¹⁵N or ¹³C

¹H-¹⁵N HSQC (Heteronuclear Single Quantum Correlation / Coherence): crosspeaks between ¹H and ¹⁵N atoms that are directly bonded

¹H-¹³C HSQC: crosspeaks between ¹H and ¹³C atoms that are directly bonded



¹H-¹⁵N HSQC spectrum of a protein with ~ 160 residues



Large proteins (or other molecules or molecular assemblies)

 ¹⁵N and ¹³C labeling combined with partial or complete ²H labeling (deuteration) to reduce relaxation



Garrett DS, Seok YJ, Liao DI, Peterkofsky A, Gronenborn AM, Clore GM. Solution structure of the 30 kDa N-terminal domain of enzyme I of the Escherichia coli phosphoenolpyruvate:sugar phosphotransferase system by multidimensional NMR. Biochemistry. 1997 Mar 4;36(9):2517-30. Large proteins (or other molecules or molecular assemblies)

TROSY (Transverse Relaxation-Optimized SpectroscopY):

- similar to HSQC
- sharper peaks for large molecules
- at high magnetic fields



Rule, G.S. and Hitchens, T.K. Fundamentals of Protein NMR Spectroscopy, 2006, Springer







HNCOCACB



Residue	C=0	Cα	C^{ℓ}	Others
Ala	177.8	52.5	19.1	
Cys (reduced)	174.6	58.2	28.0	
Cys (oxidized)	174.6	55.4	41.1	
Asp	176.3	54.2	41.1	γCO 180.0
Glu	176.6	56.6	29.9	γCH ₂ 35.6; δCO 183.4
Phe	175.8	57.7	39.6	1C 138.9; 2,6CH 131.9; 3,5CH 131.5; 4CH 129.9
Gly	174.9	45.1		
His	174.1	55.0	29.0	2CH 136.2; 4CH 120.1; 5C 131.1
Ile	176.4	61.1	38.8	γCH ₂ 27.2; γCH ₃ 17.4; δCH ₃ 12.9
Lys	176.6	56.2	33.1	γCH ₂ 24.7; δCH ₂ 29.0; εCH ₂ 41.9
Leu	177.6	55.1	42.4	γCH 26.9; δCH ₃ 24.9, 23.3
Met	176.3	55.4	32.9	γCH ₂ 32.0; εCH ₃ 16.9
Asn	175.2	53.1	38.9	γCO 177.2
Pro	177.3	63.3	32.1	γCH ₂ 27.2; δCH ₂ 49.8
Gln	176.0	55.7	29.4	γCH ₂ 33.7; δCO 180.5
Arg	176.3	56.0	30.9	γCH ₂ 27.1; δCH ₂ 43.3; εC 159.5
Ser	174.6	58.3	63.8	
Thr	174.7	61.8	69.8	γCH ₃ 21.5
Val	176.3	62.2	32.9	γCH ₃ 21.1, 20.3
Trp ^a	176.1	57.5	29.6	2CH 127.4; 3C 111.2; 4CH 122.2; 5CH 124.8; 6CH 121.0; 7CH 114.7; 8C 138.7; 9C 129.5
Tyr	175.9	57.9	38.8	1C 130.6; 2,6CH 133.3; 3,5CH 118.2; 4C 157.3

RANDOM COIL ¹³C CHEMICAL SHIFTS FOR THE 20 COMMON AMINO ACIDS WHEN FOLLOWED BY ALANINE

Wishart DS, Bigam CG, Holm A, Hodges RS, Sykes BD. 1H, 13C and 15N random coil NMR chemical shifts of the common amino acids. I. Investigations of nearest-neighbor effects. J Biomol NMR. 1995 Jan;5(1):67-81.

¹³C chemical shifts are sensitive to secondary structure \rightarrow Prediction of Φ and Ψ dihedral angles



Rule, G.S. and Hitchens, T.K. Fundamentals of Protein NMR Spectroscopy, 2006, Springer

¹³C chemical shifts can be used to estimate secondary structure propensity in intrinsically disordered proteins



Marsh JA, Singh VK, Jia Z, Forman-Kay JD. Sensitivity of secondary structure propensities to sequence differences between alpha- and gamma-synuclein: implications for fibrillation. *Protein Sci.* 2006;15(12):2795-2804.

Examples of heteronuclear 3D experiments

HNCO HN(i), N(i), C'(i-1) HN(CA)CO HN(i), N(i), C'(i)+C'(i-1)HN(i), N(i), C α (i)+C α (i-1) HNCA HN(CO)CA HN(i), N(i), C α (i-1) **HNCACB** HN(i), N(i), C α (i)+C β (i)+C α (i-1)+C β (i-1) CBCA(CO)NH HN(i), N(i), C α (i-1)+C β (i-1) C(CO)NH HN(i), N(i), all C(i-1)H(CCO)NH HN(i), N(i), all HC(i-1)CCH-TOCSY HC(i), C(i), all C(i)HCCH-TOCSY HC(i), C(i), all HC(i)15N-edited TOCSY HN(i), N(i), all H(i) 15N-edited NOESY HN(i), N(i), H(<5Å) 13C-edited NOESY HC(i), C(i), H(<5Å)

Sequential assignment using 3D heteronuclear experiments



3D ¹⁵N-edited NOESY: Distance information



3D ¹⁵N-edited NOESY: Distance information



Structure calculation:

Molecular dynamics simulation using distance (and other) restraints





Chemical Exchange



$$k_{ex} = k_1 + k_2$$

Exchange rate

 $k_{ex} \gg \Delta \nu$: fast exchange Single line: $\omega = p_A \omega_A + p_B \omega_B$

Population (fraction) of A

 $k_{ex} \approx \Delta v$: intermediate exchange broad line(s)

 $k_{ex} \ll \Delta \nu$: slow exchange Two separate lines at ω_A and ω_B Integrated intensities $\propto p_A$ and p_B

Rule, G.S. and Hitchens, T.K. Fundamentals of Protein NMR Spectroscopy, 2006, Springer



$$k_{ex} = k_{on}[L] + k_{off}$$

Exchange rate

$$K_d = \frac{k_{off}}{k_{on}} = 100 \mu M$$



Rule, G.S. and Hitchens, T.K. Fundamentals of Protein NMR Spectroscopy, 2006, Springer

Mapping ligand binding using chemical shift perturbations



Briknarová K, Zhou X, Satterthwait A, Hoyt DW, Ely KR, Huang S. Structural studies of the SET domain from RIZ1 tumor suppressor. Biochem Biophys Res Commun. 2008 Feb 15;366(3):807-13.

Hydrogen exchange

- Hydrogen atoms attached to N, O and S exchange with hydrogens in H₂O
- Hydrogens that exchange fast are not observable
- When the protein is dissolved in D₂O (heavy water) instead of H₂O, exchangeable hydrogens will be replaced by D (= deuteron = ²H)
- Exchangeable hydrogens that participate in hydrogen bonding are (somewhat) protected from exchange with solvent



Wüthrich, K. NMR of Proteins and Nucleic Acids, 1986, John Wiley & Sons